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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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Holt Modern Chemistry Chapter 7 Holt Modern Chemistry Chapter 7 The chemical formula for an ionic compound represents one formula unit - the simplest ratio of the compound's positive ions (cations) and its negative ions (anions). Monatomic cations are identified simply by the element's name.

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